Variation of Ca, Sr, Ba and Mg in the Otolith of Mudskipper in West Coast of Peninsular Malaysia

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Abstract: A study on elemental composition in the otolith of giant mudskipper, Periophthalmodon schlosseri, was done from June to October 2003. Specimens were obtained from the mangrove areas of Kuala Selangor, Sepang and Melaka in the west coast of Peninsular Malaysia. A total of 70 sagitta otoliths were analyzed to detect variation of Sr, Ba and Mg, replacing the natural chemical composition of the otolith, which is the calcium carbonate (CaCO₃). The average ratio of Sr:Ca was 0.11×10⁻⁴, Ba:Ca was 5.7×10⁻⁵ and Mg:Ca was 0.2×10⁻³. Strong correlation (R>0.8) between fish body size and otolith weight of mudskipper (p<0.01) also found during this study.

Key words: Fish, ear stone, Mangrove forest, trace element

INTRODUCTION

Otolith has been widely used as a tool in determining the age, fish growth, reproduction and fish migratory pattern (Secor and Rooker, 2000; Dwyer et al., 2003; Grandcourt et al., 2006; Walther and Thorrold, 2008; Leakey et al., 2009) in all types of fish, such as demersal and pelagic. Recent researches on otolith-water chemistry have been focused on interaction between ambient water and the otolith (Milton and Chenery, 2001). Giant mudskipper, Periophthalmodon schlosseri is different from the common fish because it spends more time on land than in water, searching, hunting, stalking and consuming prey in the mudflat (Mazlan et al., 2006). Furthermore, P. schlosseri can tolerate different types of ambient water (e.g., brackish, seawater and freshwater) and use the land adapted gill to breathe on land (Mc Inerny and Gerard, 1996). This fish holds air in its large buccopharyngeal-opercular cavity, where respiratory gas exchange takes place via gill and highly vascularized epithelial lining cavity (Aguilar et al., 2000). The entire life span of giant mudskipper is spent mainly on mudflat, where they dig burrows to seek refuge, lay eggs and escape from marine predators (Brillet, 1969; Sasekumar et al., 1994). Mudskippers generally deposit their eggs on the ceiling of the burrow and continue to develop until they are independent enough to leave the burrows (Kobayashi et al., 1971; Clayton, 1993). The embryo aerates using the oxygen stored in the burrow by the parent (Ishimatsu et al., 1998).

Otolith functions as elemental recorder, which grows throughout the lifespan of fish and the materials incorporated onto the otolith matrix become permanent, even though the fish is distressed or starved (Campana and Neilson, 1985; Miller et al., 2006). Eidsdon and Gillanders (2004) have shown that a significant relationship exist between ambient water chemistry and otolith chemistry. Most of the studies involving divalent elements, such as Mg, Ba, Mn and Sr, show strong relationship, while other elements, such as Li, Cu and Pb, do not show significant relationship (Milton and Chenery, 2001), suggesting that divalent element might substitute Ca in the calcified argonitic structure of otolith. Ba and Sr have become important tools for determining fish migratory based on the assumption of the different content of Sr and Ba in seawater, brackish water and freshwater (Eidsdon and Gillanders, 2003a, b). Compared to scales and spines, which are made from calcium phosphate, the discrete layer of otolith is made from calcium carbonate and is not directly in contact with ambient water (Gillanders, 2001). Studies on diadromous fish (e.g., sea bass) are more documented compared with those of mudskippers, which are amphidromous.

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The aim of this study is to measure the concentration of Ca, Sr, Ba and Mg in the giant mudskipper sagitta otolith and to identify the influence of the menacing environment of the unique giant mudskipper to the otolith chemistry.

**MATERIALS AND METHODS**

Three sampling locations of mudskipper habitat were selected namely Kuala Selangor, Sepang and Melaka from the west coast of Peninsular Malaysia (Fig. 1). This species is usually found and abundant in the mangroves and mudflat areas. In Melaka coast, the giant mudskippers were found on the mudflats but Kuala Selangor and Sepang areas were found in mangroves area. The salinity value at the sampling area ranged from 6-32 psu (Phua, 2004; Tee, 2003). About 70 specimens of fish sample were collected from June to September 2003 in four occasions for Kuala Selangor, two occasions each for Sepang and Melaka (Table 1). Then the fish samples were dissected to extract their otolith (Secor et al., 1991).

All glassware and plastic materials used in the laboratory and during sampling were soaked in 5% of nitric acid (HNO₃) overnight. Fish samples were randomly collected using fishing rods and nets. Specimens were weighed with electronic balance up to 0.01 g and the

<table>
<thead>
<tr>
<th>Month</th>
<th>Locality</th>
<th>Sex (F/M)</th>
<th>No. of fish</th>
</tr>
</thead>
<tbody>
<tr>
<td>June</td>
<td>Melaka</td>
<td>5/6</td>
<td>11</td>
</tr>
<tr>
<td>July</td>
<td>Kuala Selangor</td>
<td>5/7</td>
<td>12</td>
</tr>
<tr>
<td>August</td>
<td>Kuala Selangor</td>
<td>5/5</td>
<td>10</td>
</tr>
<tr>
<td></td>
<td>Melaka</td>
<td>3/2</td>
<td>5</td>
</tr>
<tr>
<td>September</td>
<td>Sepang</td>
<td>2/4</td>
<td>6</td>
</tr>
<tr>
<td></td>
<td>Kuala Selangor</td>
<td>8/6</td>
<td>14</td>
</tr>
<tr>
<td></td>
<td>Sepang</td>
<td>0/1</td>
<td>1</td>
</tr>
<tr>
<td>October</td>
<td>Kuala Selangor</td>
<td>5/6</td>
<td>11</td>
</tr>
<tr>
<td>Total</td>
<td></td>
<td>33/37</td>
<td>70</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Element</th>
<th>Analyzed value</th>
<th>Certificate value</th>
<th>Recovery (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ca</td>
<td>34.20±0.028%</td>
<td>38.80±0.5%</td>
<td>88.90±3.42</td>
</tr>
<tr>
<td>Sr</td>
<td>2.19±1.3 mg g⁻¹</td>
<td>2.16±0.05 mg g⁻¹</td>
<td>92.75±5.49</td>
</tr>
<tr>
<td>Mg</td>
<td>18.50±0.38 µg g⁻¹</td>
<td>21.00±1.00 µg g⁻¹</td>
<td>87.63±0.28</td>
</tr>
<tr>
<td>Ba</td>
<td>2.76±0.26 µg g⁻¹</td>
<td>2.89±0.08 µg g⁻¹</td>
<td>95.51±8.25</td>
</tr>
</tbody>
</table>

Fig. 1: Samplings locations in Kuala Selangor, Sepang (Selangor) and Melaka in the West coast Peninsular of Malaysia
length was measured using a measuring scale board to the nearest 0.1 cm scale. The oolith length was measured with a digital electronic calliper with accuracy ±0.01 mm while the oolith samples were weighted using the electronic balance with an accuracy of ±0.01 mg.

About 10 mL of concentrated HNO₃ was added into a glass beaker to digest the sagitta oolith. The samples were dried and redissolved with 20 mL of 0.5 M HCl. Concentrations of Ca, Sr, Ba and Mg were measured using the Perkin Elmer Optima 4300DV Inductive Coupled Plasma Omitted Emission Spectrophotometer (ICP-OES). Replicates of Certified Reference Material No. 22 (Oolith) were also prepared using the same procedure as oolith samples where the measurement results were in the range of certified values (Table 2).

RESULTS

Relationship between giant mudskipper and oolith: The shape of the extracted sagitta oolith is round (Fig. 2). The heaviest and longest fish in Melaka acquired the heaviest and longest sagitta oolith, while the smallest and shortest fish in Sepang produced the lightest oolith (Table 3, 4). The difference in weight suggest that it is as a better indicator for relative growth of giant mudskipper in coherence with higher correlation value between fish length/weight with oolith weight (R=0.89, p<0.01, Table 4). The correlation coefficient shows strong relationship between giant mudskipper and oolith (R>0.8, p<0.01) supporting the assumption that fish size (much more likely to represent the fish growth) influences the size of oolith (Table 4). This indicates that oolith development of giant mudskipper is dependent on cell growth. The relationship between fish length-weight revealed the highest correlation (R = 0.979, p<0.01), supporting that fish length various with length (Table 4).

The strong relationship of the oolith and fish suggests that oolith might be a useful indicator of the size of a giant mudskipper (Table 4).

**Elemental analysis of oolith:** The concentration pattern of the elements varied from 64.30 to 198.69 mg g⁻¹ for Ca, 0.29 to 10.86 mg g⁻¹ for Sr, 0 to 31.49 μg g⁻¹ for Ba, 11.19 to 575.14 μg g⁻¹ for Mg. There was no clear trend for average concentration of Ca and Sr obtained during this study (Fig. 3a-d). In comparison between the analyzed

<table>
<thead>
<tr>
<th>Localities</th>
<th>Total length (cm)</th>
<th>Body weight (g)</th>
<th>Oolith length (mm)</th>
<th>Oolith weight (mg)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Kuala Selangor</td>
<td>18.86±2.69</td>
<td>70.34±38.09</td>
<td>5.60±0.83</td>
<td>27.13±4.65</td>
</tr>
<tr>
<td>Melaka</td>
<td>22.39±2.32</td>
<td>124.34±45.47</td>
<td>6.85±1.00</td>
<td>49.00±2.20</td>
</tr>
<tr>
<td>Sepang</td>
<td>18.86±2.21</td>
<td>66.00±21.06</td>
<td>5.79±0.79</td>
<td>23.32±3.80</td>
</tr>
<tr>
<td>Mean</td>
<td>13.54±3.49</td>
<td>63.41±39.64</td>
<td>5.56±2.87</td>
<td>26.17±3.74</td>
</tr>
</tbody>
</table>

Fig. 2: A pair of giant mudskipper’s sagitta.

Fig. 3: Average concentrations of (a) Ca, (b) Sr, (c) Ba and (d) Mg in the giant mudskipper oolith according to months.
Fig. 4: Graph of (a) Sr:Ca, (b) Ba:Ca and (c) Mg:Ca ratios (Mean±SE) in the giant mudskipper otolith according to months

| Table 4: Relationship between giant mudskipper and otolith, Pearson correlation |
|-----------------------------|-----------------------------|-----------------------------|-----------------------------|
| Otolith weight              | Otolith total length        | Giant mudskipper total length | Giant mudskipper weight     |
| Otolith weight              | 1                           | 0.886**                    | 0.889**                     | 0.903**                  |
| Otolith total length        | 0.885**                    | 1                           | 0.820**                     | 0.832**                  |
| Giant mudskipper total length | 0.889**                    | 0.820**                     | 1                           | 0.979**                  |
| Giant mudskipper weight     | 0.903**                    | 0.832**                     | 0.979**                     | 1                        |

n = 70, **Correlation is significant at the 0.01 level (2-tailed), p<0.01

elements, Ba recorded concentration. In Fig. 3c and d, average concentrations of Ba and Mg showed inverse relationship, where the lowest and highest Ba concentration was recorded in June and September and in reverse with, respectively.

From this results, the monthly distribution pattern of Sr:Ca ratio and Sr average concentration differed, in which the highest Sr concentration was observed in September, while Sr:Ca ratio was maximum in October (Fig. 3b, 4a). Ratios of Ba:Ca and Mg:Ca revealed a similar trends with Ba and Mg average concentration (Fig. 3c, d, 4b, c).

**DISCUSSION**

**Characteristic of Ca, Sr, Ba and Mg in the mudskipper otolith:** Otolith was developed from the daily crystallization process of fluid within the endolympathic canal of the fish inner ear in the form of concentric layers series around the nucleus (Farrell and Campana, 1996; Thorrold et al., 1997). This growth layer will stop when the growth of the fish stops which is directly related to otolith size/weight to fish size/weight (Romanek and Gauldie, 1996).

Mazlan and Rohaya (2008) also found high correlation (R>0.9) between daily rings of otolith with the size of giant mudskipper, similar to the results obtained from this research. The variability of the otolith shape decreases the accuracy of the length measured, compared to otolith weight (Strelcheck et al., 2003). In addition, otolith weight is a preferred indicator of relative growth difference in juvenile gug (Mycteroperca microlepis) compared to otolith length (Strelcheck et al., 2003). Mazlan and Rohaya (2008) estimated that the growth parameter for giant mudskipper were L₅₀ = 29 cm and K = 1.4 year⁻¹, where the body growth is influenced by the growth of otolith.

Generally, otolith is composed of 99% CaCO₃ (Payan et al., 2004) and Ca is the highest element in otolith (Fig. 3a). The presence of Sr, Mg and Ba by ranking in the otolith suggest that other divalent transition element can substitute Ca because the atomic radii of Mg, Ba and Sr are 0.065, 0.113 and 0.135 Å, respectively, similar to the Ca atomic radius, 0.099 Å (Gauldie et al., 1995). The other elements such as Sr, Mg and Ba have similar atomic radii with Ca in the ambient water are taken in through the gill by active transport of Ca. The Ca transport system, located in chloride cells of gills (Campagna, 1999; Chowdhury and Bust, 2002), takes up Sr greater than Ba because concentration of Sr in brackish water exceeds that of Ba (Hamer et al., 2006) (Fig. 3b). In contrast, the concentration of Mg in brackish water is higher than Sr but the uptake of Sr is greater than Mg (Hamer et al., 2006). It also means that the concentration levels of trace elements in otolith are not directly related to the level of trace elements in water column at study site.
Despite other calcified structures, such as corals and bones, the incorporation of elements onto otolith consists of three main interfaces: brachial uptake, cellular transport and crystallization. Elements, which are mostly divalent elements, must pass these barriers before they are crystallized onto otoliths matrix (Campagna, 1999). In fact, the final precipitation of elements onto otolith is due to physical chemistry of biomineralisation (Nielsen and Christoffersen, 1982), which is affected by physiological process (Ca') and environmental variables (Hoff and Fuiman, 1995). The elements incorporate onto otolith which represents the influence of the environment to the entire life span of the giant mudskipper, which inhabits the mangrove throughout its lifetime (Ishimatsu et al., 1998).

Salinity recorded in study sites is between 6-32 psu (Phua, 2004; Tee, 2003), which was lower than that of seawater. Water flow in the study areas was affected by seawater intrusion from the tidal wave in the mangrove systems and dilution was influenced by the freshwater input from the nearest river (Pritchard, 1955). Hence, the concentration level of elements is increased due to the mixing of freshwater and seawater happens. In the present study, Sr concentration is high compared to freshwater because of the high salinity (Fig. 3). Even though the monsoon season brought heavy rainfall from September to December, no significant difference is traced from Sr, Mg and Ba, trends from September to October (Fig. 3b, c). There were also no obvious trends indicating the differences from the pre-monsoon season suggesting that monsoon season is not affecting the water flow system in the study areas (Fig. 3). The monthly Mean Sea Level (MSL) recorded in Melaka and Selangor was insignificantly different (JUPEM, 2003).

Brackish water fishes are usually high uptake of Sr than Ba because concentration level of Sr in brackish water column (de Vries et al., 2005). While, Mg and Ba contents show the contradictory pattern suggesting that Mg and Ba are replacing each other by getting the influence of the freshwater input in the study area (Fig. 3c, d).

Giant mudskippers inhabit estuary in the tidal mudflat in the mangroves forest. Generally, the hydrology system in mangroves of Malaysia is influenced by rainfall, evaporation and tidal flushing (Chan, 1987). It has been proven that Ba content is higher in ambient water with fresh water intrusion, which is rain water and river (Elsdon and Gillander, 2003a, b, 2004). This freshwater intrusion factors most likely is influence by monsoon season and tides. However, this study revealed that these factors were not significantly affect the Ba content in otolith (Fig. 3c). During monsoon season, heavy rain fall will happen from September to December and less rain during pre-monsoon period (Cheang, 1987). This will decrease the salinity value of ambient water and enhance the Ba levels. In this study, suggesting, the water contents of estuary will affect the concentration levels of Ba in the otolith (Fig. 3c). This also indicates, the sampling stations located in mangrove area is real estuarine (Pritchard, 1955), where a low salinity value at the top layer and high value at the bottom layer of water body.

In addition, larvae of giant mudskippers develop in deep burrows dug in soft sediment in mudflats (Brillet, 1969). Although burrow condition is hypoxic, the larvae can still survive because the burrows accumulate with air by the adult not only for the larvae to develop, but also to provide oxygen to adult during thermoregulate (Tytler and Vaughan, 1983) and avoiding from predators (Sasekumar et al., 1994). The giant mudskipper acquires low salinity water from the surface of the water body and did not swim deeper or stay in the high salinity of water body. This study suggests that high Ba and Sr in the otolith suggested that the water body inhabited by the giant mudskipper was brackish water and low Mg support the low sea water intrusion to the study area.

### Ratios of elements to calcium

Calcium (Ca) is locally incorporated in otolith but others divalent elements such as Sr, Ba and Mg, have the ability to substitute with calcium ion in natural chemical composition depending on environmental factors (Gauldie et al., 1995). The ratio of elements to Ca is determined to detect the variation of substitute elements (i.e., Sr, Ba and Mg) in the otolith. The ratios of elements to Ca are lower than the values of seawater, brackish water and freshwater (Table 5). The incorporation of Ca on otolith matrix is a natural process and not related to the content of Ca in the ambient water (Farrell and Campana, 1996, Thorrold et al., 1997).

In this study, the ratios of Sr:Ca are used as indicators to investigate the changing of salinity in various marine environments (Campana and Tzeng, 2000) because high Sr:Ca ratio value is found in the saline waters (Table 5). On the other hand, Mg:Ca ratios were used as elemental fingerprints (Campana and Tzeng, 2000) but the Ba:Ca ratios will reveal the migration patterns of fish from brackish water to low salinity gradient value.

<table>
<thead>
<tr>
<th>Ratio</th>
<th>Sea water</th>
<th>Brackish water</th>
<th>Fresh water</th>
<th>Otolith</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sr:Ca</td>
<td>1.9x10^-2</td>
<td>1.7x10^-2</td>
<td>1.0x10^-2</td>
<td>5.60±4.5x10^-3</td>
</tr>
<tr>
<td>Ba:Ca</td>
<td>3.0x10^-3</td>
<td>1.8x10^-3</td>
<td>1.7x10^-2</td>
<td>1.20±1.5x10^-3</td>
</tr>
<tr>
<td>Mg:Ca</td>
<td>3.13</td>
<td>3.07</td>
<td>0.25</td>
<td>1.67±9.0x10^-4</td>
</tr>
</tbody>
</table>

*Summerhayes and Thorpe (1996), **Hamer et al. (2006), ***SRM NIST 1643e, ****Giant mudskipper otolith ratios
during breeding season (Milton and Chenery, 2001). But during investigation, the giant mudskippers were staying near to their burrows with a low freshwater input. The foreign divalent element (e.g., Sr, Ba and Mg) ratio to Ca indicates the acceptance of the otolith towards these elements based on the similarity of the atomic radii to Ca atomic radius. The Sr:Ca ratios were the highest ratio obtained during this study because of its availability in ambient water. While the abundant of Mg value in otolith is not only as signal the seawater input but also indicate the uptake process occur during fish growing in various marine environment.

This research also revealed that the dominance life time on giant mudskippers were on mudflat by comparing with the Sr:Ca, Ba:Ca and Mg:Ca ratios in seawater, brackish water and freshwater system. Then during investigation the giant mudskippers are moving in the ranged of 4200 square feet from their burrow for food searching (Murdy, 1986) as shows by the low ratio value of Ba:Ca from June to October 2003 (Fig. 4b). The contradictory pattern of Mg:Ca ratio and Bar:Ca ratio found in the otolith is similar ratios in seawater, brackish water and freshwater (Table 3) but that does not mean that Ca in seawater affected otolith. Ca incorporated on in otolith by the physiological endolympah uptake of fish (Simkiss, 1974; Farrell and Campana, 1996).

CONCLUSION

The concentration levels of Ca, Sr, Ba, and Mg in otolith are related to the ambient water inhabited by giant mudskippers will revealed as Sr>Mg>Ba. The suitability and preferences of elements not only depends on the atomic radius closest to Ca atomic radius, but also the abundances of divalent elements (Mg and Ba) in the ambient water causing high uptake of elements onto otolith. The ratio of Sr:Ca is higher than Ba:Ca ratio, indicating that the lifespan of giant mudskipper is spent near the burrow area, they do not migrate to less saline (<6 psu) water body, upstream nor downstream.

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REFERENCES


Mc Inerny and G. Gerard, 1996. All About Tropical Fish. 3rd Edn., Jarrold and Sons Ltd., Norwich.


